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Research Article

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## Quantum Molecular Modeling of Piperidine-4-Carboxamide Derivative CCR5 Antagonist (TAK-220) With Anti-HIV-1 Activity

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### ABSTRACT

QSAR and SAR studies on the Piperidine-4-Carboxamide Derivatives as non- nucleotide reverse transcriptase inhibitor of HIV-1 using the topological, physicochemical, and hydrophobic parameters, indicator parameters along with the quantum parameters. Application of multiple linear regression analysis indicated that a combination of different molecular descriptors and the indicator parameters yielded a statistically significant model for the prediction of activity,  $CCR5^{50} \log IC_{50}$ . The final selection of a potential Piperidine-4-Carboxamide Derivatives as non- nucleotide reverse transcriptase inhibitor of HIV-1 is made by the quantum molecular modeling.

**Keywords:** QSAR, Anti HIV-1, Topological indices, physicochemical properties and quantum descriptors and  $\log IC_{50}$ .

### ARTICLE INFO

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### 1. Introduction

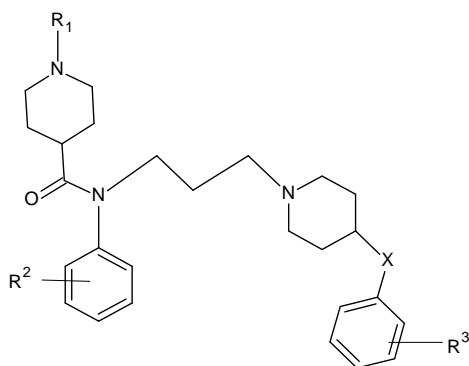
Despite worldwide efforts to prevent the spread of human immunodeficiency virus type 1 (HIV-1), the number of HIV-1-infected people still continues to rise.<sup>1</sup>Recent International Journal of Chemistry and Pharmaceutical Sciences

advances of chemokine receptors functioning as HIV-1 co-receptors have provided a novel strategy for controlling HIV-1 infection.<sup>2</sup> HIV-1 strains that cause the initial

infection primarily utilize CC chemokine receptor 5 (CCR5),<sup>3</sup> and CCR5-using (R5) HIV-1 is isolated predominantly during the asymptomatic stage of the infection, which usually persists for 5-10 years.<sup>4</sup> CCR5 belongs to the seven-transmembrane G protein-coupled receptor super family and its natural ligands include the CC chemokines [regulated on activation, normal T cell expressed, secreted (RANTES), macrophage inflammatory protein (MIP)-1R, and MIP-1 $\alpha$ ], which have been reported to inhibit R5 HIV-1 infection in vitro.<sup>5</sup> Subsequent optimization identified a series of piperidine-4-carboxamide derivatives, exemplified low nanomolar affinity for CCR5 and exhibited good anti-HIV-1 activity.<sup>6</sup> In vitro metabolic stability studies in human hepatic microsomes, however, showed these compounds to be rapidly metabolized.<sup>7</sup>

## 2. Materials and Methods

Quantitative structure-activity relationships (QSAR) have been established for set of 21 analogues of Piperidine-4-Carboxamide (fig.1) a potent inhibitor of the HIV-1 reverse transcriptase (RT). The activity of these compounds was adopted from the literature .Table 1.



**Figure 1:** Parent structure of Piperidine-4-carboxamide Derivatives used in present study

### Quantum Descriptors

In this category following descriptors were tested, these descriptors represents different electronic environment of the compound.

**Total energy:** The total energy serves to represents the chemical reactivity of the compounds.

**Binding Energy:** It is used as a measure of electronic interaction between drug and receptor

**HOMO (Highest occupied Molecular Orbital) & LUMO (Lowest Unoccupied Molecular Orbital):**

Energies associated with the highest occupied molecular orbital (EHOMO), and lowest unoccupied molecular orbital (ELUMO) are often good candidates for 2-dimensional descriptors. The HOMO energy is related to the ionization potential and is a measure of the molecule's tendency to be attacked by electrophiles. Correspondingly, the LUMO energy is related to the electron affinity and is a measure of a molecule's tendency to be attacked by nucleophiles.<sup>8,9</sup>

**Dipole Moment:**

The polarity is represented by the dipole moment. Polaris ability is a tensor relating the induced dipole moment to the applied electric field strength.

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### X, Y, Z coordinate of N atom:

Represents the electron density in all the 3 dimension of the molecule, but here Nitrogen atom has been selected , because it is an atom where regular substitution are observed.

### Net charge and Electron density of few selective atoms.

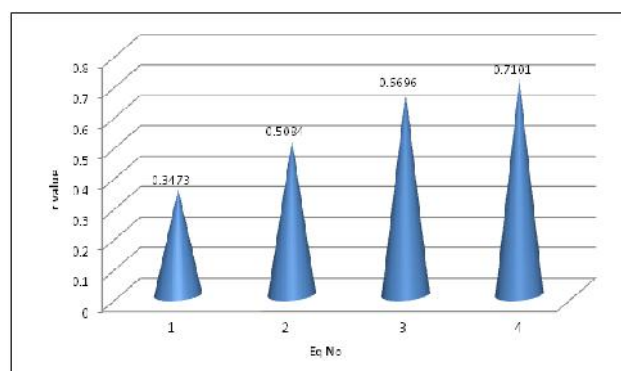
Electron densities and net Charges on atoms are considered as non directional reactivity indices.

## 3. Results and Discussion

As mentioned in introduction QSAR studies on set of Piperidine-4-Carboxamide derivative containing 21 compounds is performed. The Quantum Descriptors tested in present study has been given in **Table 2 a, b & c**. The QSAR analysis using stepwise multiple linear regression method is also performed separately with Quantum descriptors, in a same manner as with topological descriptors and physicochemical descriptors. Here quantum descriptors are used as independent variable to predict the biological activity (dependent variable).

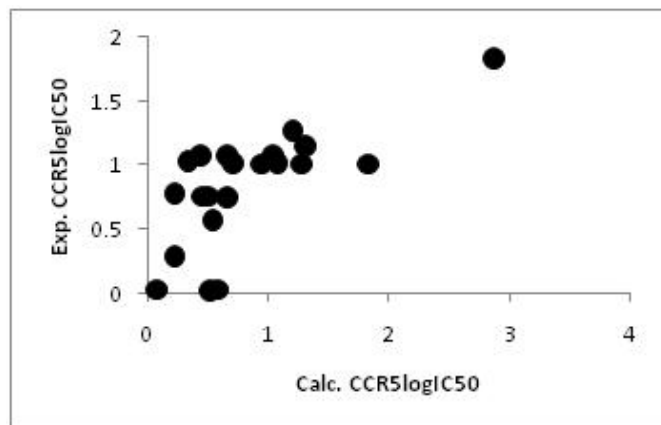
The correlation matrix is shown in the form of **Table 3** there are two descriptors showing approximately same correlation with biological activity. Therefore while testing bivariate combination, both the descriptors are examine with the second descriptor, or in words, in first step of screening two descriptors are screened. These are Ims and EDN (Electron density at Nitrogen of R<sub>1</sub>) with correlation value 0.34727 and -0.36306 respectively. The relative raise in the r value can be easily observed by demonstrating the graph below as **Figure 2**. It is clearly observed from the model obtained from quantum descriptors that EDN produces no significant result in further steps of regression analysis, therefore screened out from the regression analysis, and finally no appear in the final mathematical model.

The best correlation obtained using Quantum descriptors is 0.7101, this result is a significant result, and the model with such result may consider as the an efficient model for the prediction purpose, also it is worthy for the screening of HOMO and EDC4 from the other Quantum descriptors. These two descriptors are not seems as an important descriptor initially in correlation matrix, but both the descriptors are comeout as an operational descriptors.



**Figure 2 :** Relative raise in the value of “r” from Eq (1) to Eq (4)

By the persual of Eq (4) *I<sub>ms</sub>*, *I<sub>x</sub>*, HOMO and EDC4 are selected as a predictive descriptors among the physicochemical descriptors. The prediction of Biological activity i.e., inhibition of 125I-labeled RANTES binding to Chinese hamster ovary (CHO) cells expressing human CCR5 on their surface is calculated or predicted using mathematical model obtained from quantum descriptors and present in **Table 4**. The correlation between experimental and calculated activity is present in **Figure 3**



**Figure 3:** Graph between Experimental and Calculate biological activity using Eq (4)

#### 4. Conclusion

The negative coefficient of *I<sub>x</sub>* indicate that presence of –CH<sub>2</sub>– group decreases value of biological activity in a quantitative manner, i.e., presence of –CH<sub>2</sub>– is favorable. On the other hand positive coefficient of *I<sub>ms</sub>*, leads to the increase in the value of biological activity in a quantitative aspects, i.e., presence of mesityl group in unfavorable. The positive coefficient of HOMO shows that the value of **CCR<sub>5</sub>logIC<sub>50</sub>** increases with the increase in the value of HOMO, higher value of CCR<sub>5</sub>logIC<sub>50</sub> is unfavorable. Therefore molecule should have such structure which possesses lower value of HOMO. Similarly The positive coefficient of EDC4 shows that the value of CCR<sub>5</sub>logIC<sub>50</sub> increases with the increase in the value of EDC4 (Electron density at 4<sup>th</sup> Carbon), higher value of CCR<sub>5</sub>logIC<sub>50</sub> is unfavorable. **HOMO > EDC4 > I<sub>ms</sub> > I<sub>x</sub>**. The QSAR model in the form of Eq (4) is the model obtained from Quantum parameters, it contains HOMO and EDC4 with the positive coefficient, and it means both the parameters are directly proportional to the value of CCR<sub>5</sub>logIC<sub>50</sub>. That means higher value of HOMO and EDC4 raise the value of CCR<sub>5</sub>logIC<sub>50</sub>, which implies that the higher concentration is needed for 50% inhibition activity. Therefore substitution which lower the energy of HOMO and Electron density at C4 is need to substitute in the compound to make is more biologically active.

**Table 1:** Biological activity and different substituent of Piperidine-4-carboxamide

Compound	R <sup>1</sup>	R <sup>2</sup>	X	R <sup>3</sup>	CCR <sub>5</sub> <sup>a</sup> logIC <sub>50</sub> (nm)
1	Ms	3,4-diCl	CH <sub>2</sub>	4-MS	0.342
2	Ms	3,4-diCl	CH <sub>2</sub>	4-F	0.519
3	Ac	3,4-diCl	CH <sub>2</sub>	4-F	0.079
4	Ac	H	CH <sub>2</sub>	H	1.204
5	Ac	3,4-diCl	S	4-F	0.23
6	Ac	3,4-diCl	SO	4-F	0.505
7	Ac	3,4-diCl	SO <sub>2</sub>	4-F	0.462
8	Ac	3,4-diCl	NH	4-F	1.301
9	Ms	3,4-diCl	NHSO <sub>2</sub>	4-F	0.662
10	Ms	3,4-diCl	NHCO	4-F	2.863
11	Ms	3,4-diCl	CH <sub>2</sub>	4-CN	0.23
12	Ms	3,4-diCl	CH <sub>2</sub>	4-CO <sub>2</sub> Me	0.663
13	Ms	3,4-diCl	CH <sub>2</sub>	4-CO <sub>2</sub> H	1.82
14	Ms	3,4-diCl	CH <sub>2</sub>	4-CONH <sub>2</sub>	0.949
15	Ms	3,4-diCl	CH <sub>2</sub>	3- CONH <sub>2</sub>	0.447
16	Ms	3,4-diCl	CH <sub>2</sub>	2- CONH <sub>2</sub>	1.079
17	Ms	3,4-diCl	CH <sub>2</sub>	4-CONHMe	0.708
18	Ms	3,4-diCl	CH <sub>2</sub>	4-CONHt-Bu	1.041
19	Ms	3,4-diCl	CH <sub>2</sub>	4-CONMe <sub>2</sub>	1.279
20	Ac	3,4-diCl	CH <sub>2</sub>	4- CONH <sub>2</sub>	0.58
21	Ac	3-Cl,4-Me	CH <sub>2</sub>	4- CONH <sub>2</sub>	0.544

<sup>a</sup> Inhibition of 125I-labeled RANTES binding to CCR5-expressing CHO cells.

**Table 2a:** The Quantum Descriptors tested in present study

Compound	TE	BE	HOMO	LUMO	DM	N <sub>x</sub>	N <sub>y</sub>	N <sub>z</sub>
1	2425.7	2806.6	-0.218	-0.14	7.31	-8.70	0.805	-0.008
2	2170.4	2523.8	-0.219	-0.14	8.50	-8.18	1.59	0.19
3	2042.0	2372.9	-0.218	-0.13	11.45	-7.94	1.27	0.16

4	1788.7	2061.8	-0.218	-0.04	5.82	-7.49	0.36	0.15
5	2025.9	2360.0	-0.216	-0.13	16.80	-8.03	1.30	0.14
6	2130.5	2482.7	-0.218	-0.13	17.98	-8.67	1.33	0.18
7	2241.8	2612.1	-0.218	-0.14	15.88	-7.31	-1.45	-3.02
8	2070.3	2405.4	-0.183	-0.13	10.94	-7.78	-0.41	0.57
9	2542.7	2947.4	-0.219	-0.17	10.47	-5.57	2.63	0.19
10	2383.2	2765.2	-0.216	-0.14	13.47	-7.44	0.88	0.08
11	2371.2	2714.4	-0.122	-0.11	26.28	-5.70	0.89	-1.1E-07
12	2381.2	2757.5	-0.219	-0.14	5.67	-9.60	1.50	-0.006
13	2300.2	2669.1	-0.220	-0.14	5.99	-9.01	1.48	-0.18
14	2300.2	2663.5	-0.220	-0.14	3.07	-9.06	1.48	-0.08
15	2305.7	2668.2	-0.219	-0.14	6.51	-8.61	1.60	-0.05
16	2344.5	2707.0	-0.219	-0.14	11.22	-7.76	1.59	-0.06
17	2379.3	2749.2	-0.219	-0.14	2.85	-9.67	1.57	0.04
18	2471.8	2849.1	-0.219	-0.14	3.80	-9.87	1.47	0.01
19	2471.8	2849.1	-0.220	-0.14	3.80	-9.68	1.53	-0.04
20	2168.1	2508.0	-0.218	-0.14	7.65	-9.02	1.19	-0.12
21	2176.6	2508.5	-0.213	-0.12	7.29	-9.01	1.04	-0.12

Where, TE = Total Energy, BE = Binding energy, HOMO = Energy of highest occupied molecular orbital LUMO = Energy of lowest unoccupied molecular orbital, DM = Dipole moment in ab initio molecular field, Nx = energy at x coordinate of N atom, Ny = energy at y coordinate of N atom, Nz = energy at z coordinate of N atom

**Table 2b:** Quantum Descriptors (Net charges)

Compound	NCN	NCC3	NCC4	NCCX	NCC`X	NCCR3
1	0.10	0.24	0.21	0.16	0.14	0.10
2	0.10	0.24	0.21	0.16	0.04	0.05
3	0.02	0.24	0.21	0.16	0.04	0.05
4	2.32	0.01	2.18	0.16	0.08	2.09
5	0.02	0.24	0.21	0.17	5.52	-0.03
6	0.02	0.24	0.21	0.18	0.02	-0.04
7	0.02	0.24	0.21	0.06	0.09	-0.03
8	0.02	0.24	0.21	0.15	0.18	-0.04
9	0.04	0.24	0.21	0.27	0.09	-0.03
10	0.11	0.24	0.21	0.15	0.07	-0.05
11	-0.28	0.24	0.21	0.28	0.02	0.11
12	0.11	0.24	0.21	0.11	3.96	3.23
13	0.11	0.24	0.21	0.10	3.47	0.02
14	0.11	0.24	0.21	0.10	0.03	0.03
15	0.10	0.24	0.21	0.10	0.01	0.02
16	0.11	0.24	0.21	0.097	0.03	0.03
17	0.10	0.24	0.21	0.10	0.02	0.03
18	0.10	0.24	0.21	0.09	0.03	0.02
19	0.10	0.24	0.21	0.11	0.02	0.03
20	0.028	0.24	0.21	0.11	0.02	0.03
21	0.031	0.17	0.12	0.11	0.02	0.03

Where, NCN = Net charge at Nitrogen of R<sub>1</sub>, NCC3 = Net charge of C<sub>3</sub>, NCC4 = Net charge on C<sub>4</sub>, NCCX = Net charge on Carbon left to -X-, NCC`X = Net charge on Carbon right to -X-, NCCR3 = Net charge on Carbon of R<sub>3</sub>

**Table 2c:** Quantum Descriptors (Electron Density)

Compound	EDN	EDC3	EDC4	EDCX	EDC`X	EDCR3
1	4.89	3.76	3.79	3.84	3.86	3.90
2	4.89	3.76	3.79	3.84	3.96	4.05
3	4.98	3.76	3.79	3.84	3.96	4.05
4	4.98	3.98	4.02	3.84	3.92	3.98

5	4.98	3.76	3.79	3.84	3.96	4.05
6	4.97	3.76	3.79	3.82	3.98	4.04
7	4.97	3.76	3.79	3.94	3.91	4.04
8	4.97	3.76	3.79	3.84	3.82	4.05
9	4.95	3.76	3.79	3.73	3.90	4.03
10	4.89	3.76	3.80	3.85	3.99	4.06
11	5.28	3.96	3.45	3.72	3.97	3.89
12	4.89	3.76	3.80	3.89	3.96	3.97
13	4.89	3.76	3.79	3.90	3.97	3.98
14	4.89	3.76	3.79	3.90	3.97	3.97
15	4.90	3.76	3.80	3.90	3.98	3.97
16	4.89	3.76	3.79	3.90	3.97	3.97
17	4.89	3.76	3.79	3.90	3.97	3.97
18	4.90	3.76	3.80	3.90	3.97	3.97
19	4.89	3.76	3.79	3.89	3.97	3.97
20	4.97	3.76	3.79	3.89	3.97	3.97
21	4.97	3.83	3.88	3.89	3.97	3.48

Where, EDN = Electron Density on Nitrogen of R<sub>1</sub>, EDC<sub>3</sub> = Electron Density on C<sub>3</sub>, EDC<sub>4</sub> = Electron Density on C<sub>4</sub>, EDC<sub>X</sub> = Electron Density on Carbon left to -X-, EDC<sup>∧</sup>X = Electron Density on Carbon right to -X-, EDC<sub>R3</sub> = Electron Density on Carbon of R<sub>3</sub>

**Table 3:** Correlation matrix of Quantum descriptors with biological activity

	CCR <sub>5</sub> logIC <sub>50</sub>	TE	BE	HOMO	LUMO	IMS	IX	NCN
CCR <sub>5</sub> logIC <sub>50</sub>	1.000							
TE	.160	1.000						
BE	.163	.998	1.000					
HOMO	-.152	.022	-.005	1.000				
LUMO	.064	-.634	-.665	-.222	1.000			
IMS	.347	.673	.656	.090	-.114	1.000		
IX	-.174	.103	.070	-.008	.296	.452	1.000	
NCN	.196	-.561	-.576	-.215	.826	-.199	.155	1.000
NCC <sub>3</sub>	-.097	.606	.629	.064	-.888	.298	-.180	-.934
NCC <sub>4</sub>	.138	-.586	-.606	-.062	.874	-.223	.134	.984
EDN	-.363	-.198	-.225	.885	.331	-.210	-.080	-.076
EDC <sub>3</sub>	-.073	-.376	-.416	.587	.848	-.086	.236	.586
EDC <sub>4</sub>	.236	-.432	-.421	-.779	.304	-.322	-.004	.670
EDC <sub>X</sub>	.187	-.013	-.006	-.591	-.037	.231	.275	.025
EDC <sup>∧</sup> X	.094	.136	.127	-.130	-.000	.317	.335	-.147
	NCC <sub>3</sub>	NCC <sub>4</sub>	EDN	EDC <sub>3</sub>	EDC <sub>4</sub>	EDC <sub>X</sub>	EDC <sup>∧</sup> X	
NCC <sub>3</sub>	1.000	1.000						
NCC <sub>4</sub>	-.937	.090						
EDN	-.112	.694	1.000					
EDC <sub>3</sub>	-.732	.546	.692	1.000				
EDC <sub>4</sub>	-.606	-.088	-.647	-.096	1.000			
EDC <sub>X</sub>	.042	-.088	-.637	-.431	.452	1.000		
EDC <sup>∧</sup> X	.114	-.158	-.033	-.013	-.133	.223	1.000	

**Table 4:** Experimental and predicted biological activity

Compound	Experimental CCR <sub>5</sub> logIC <sub>50</sub>	Calculated CCR <sub>5</sub> logIC <sub>50</sub>	Residual
1	0.342	1.03	-0.6876
2	0.519	0.014	0.5053
3	0.079	0.025	0.0539
4	1.204	1.268	-0.0639
5	0.23	0.778	-0.5483
6	0.505	0.756	-0.2506

7	0.462	0.756	-0.2936
8	1.301	1.152	0.1486
9	0.662	0.744	-0.0823
10	2.863	1.837	1.0262
11	0.23	0.281	-0.0508
12	0.663	1.072	-0.4093
13	1.82	1.007	0.8131
14	0.949	1.007	-0.0579
15	0.447	1.072	-0.6253
16	1.079	1.018	0.0608
17	0.708	1.018	-0.3102
18	1.041	1.072	-0.0313
19	1.279	1.007	0.2721
20	0.58	0.025	0.5549
21	0.544	0.568	-0.0241

### Univariate model

There are two univariate model found with approximately results, therefore given in the form of Eq (1a) and (1b)

$$CCR_5 \log IC_{50} = 0.4297 (\pm 0.2662) Ims + 0.6086 \quad \text{Eq (1a)}$$

N = 21, r = 0.3473 Se = 0.6092 F = 2.606

$$CCR_5 \log IC_{50} = -2.6514 (\pm 1.5613) EDN + 0.5053 \quad \text{Eq (1b)}$$

N = 21, r = -0.363 Se = 0.6053 F = 2.884

### Bivariate model

$$CCR_5 \log IC_{50} = 0.6627 (\pm 0.2816) Ims - 0.5696 (\pm 0.3113) Ix + 0.8934 \quad \text{Eq (2)}$$

N = 21, r = 0.5084, Se = 0.5747 F = 3.138

### Trivariate model

$$CCR_5 \log IC_{50} = 0.895 (\pm 0.268) Ims - 0.683 (\pm 0.28) Ix + 3.14 (\pm 1.3) EDC4 - 11.0608 \quad \text{Eq (3)}$$

N = 21, r = 0.6696, Se = 0.5101, F = 4.606

### Tetrivariate model

$$CCR_5 \log IC_{50} = 11.34 (\pm 8.45) HOMO + 1.004 (\pm 0.274) Ims - 0.7305 (\pm 0.276) Ix + 5.403 (\pm 2.11) EDC4 - 17.25 \quad \text{Eq (4)}$$

N = 21 r = 0.7101, Se = 0.4985, F = 4.068

## 5. Acknowledgment

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